Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.003 Å R factor = 0.043 wR factor = 0.106 Data-to-parameter ratio = 9.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The structure of the title compound, $C_{17}H_{20}O_{11}$, has been determined by single-crystal X-ray diffraction at 100 K. The glucopyranose ring exhibits a chair conformation, with all non-H-atom substituents in equatorial positions, and the triple bond is sterically easily accessible, allowing for followup cycloaddition reactions. The glucopyranose ring exhibits chair conformation with all non-H substituents in equatorial positions, and the triple bond is sterically easily accessible, allowing for followup cycloaddition reactions.

Comment

glucopyranuronate

Prop-2-ynyl 1,2,3,4-tetra-*O*-acetyl-β-D-glucopyranuronate, (I), was prepared as part of a study on the synthesis of 1,2,3triazole-linked sugars. In this study, various alkynes were coupled with a D-glucosyl azide in the presence of Cu^I to form 1,4-disubstituted 1,2,3-triazoles. This reaction was found to be regiospecific, and proceeds with retention of the β-D-glucoanomeric stereochemistry (Akula *et al.*, 2004). This rapid heterocyle-forming reaction fits well the criteria set out for processes useful in the so-called 'click chemistry' (Kolb *et al.*, 2001).



The molecular structure of (I) is shown in Fig. 1. The bond lengths and angles are within expected ranges for sugars. The six-membered ring exhibits the chair conformation expected for a glucopyranose, and all non-H substituents are found in equatorial positions. The acetate groups are all twisted to be roughly perpendicular to the equatorial plane and their C=O groups point alternately up and down (see Fig. 1).

The propynyl functional group is close to linear $[C15-C16-C17 = 177.8 (3)^{\circ}]$ and the C16-C17-H17 angle is 176 (2)°. The C=C triple bond distance is 1.177 (4) Å. Steric crowding around the triple bond is not very pronounced, and both the acidic H atom and the π -electron density of the alkyne are easily accessible from several sides of the molecule. Thus, the regioselectivity of the Cu-catalysed triazole-formation reaction is best explained by electronic effects induced by the copper catalyst itself, on either an intermediate or the transition state of the cycloaddition reaction.

Experimental

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved Prop-2-ynyl 1,2,3,4-tetra-O-acetyl- β -D-glucopyranuronate was prepared as described in the literature (Akula *et al.*, 2004). Crystals

Received 14 October 2004 Accepted 4 November 2004 Online 13 November 2004

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suitable for single-crystal X-ray diffraction were obtained by slow cooling of a hot methanol solution.

Crystal data

 $C_{17}H_{20}O_{11}$ $M_r = 400.33$ Orthorhombic, $P2_12_12_1$ a = 7.3827 (9) Å b = 14.1462 (17) Å c = 18.077 (2) Å V = 1887.9 (4) Å³ Z = 4 $D_x = 1.408$ Mg m⁻³ Data collection

Bruker SMART APEX CCD diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS* in *SAINT-Plus*; Bruker, 2003) $T_{min} = 0.884, T_{max} = 0.97$ 20 026 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.043$ $wR(F^2) = 0.106$ S = 1.192663 reflections 282 parameters H atoms treated by a mixture of independent and constrained refinement Mo $K\alpha$ radiation Cell parameters from 6910 reflections $\theta = 2.9-31.8^{\circ}$ $\mu = 0.12 \text{ mm}^{-1}$ T = 100 (2) KBlock, colourless $0.37 \times 0.22 \times 0.22 \text{ mm}$

2663 independent reflections 2634 reflections with $I > 2\sigma(I)$ $R_{int} = 0.021$ $\theta_{max} = 28.3^{\circ}$ $h = -9 \rightarrow 9$ $k = -18 \rightarrow 18$ $l = -23 \rightarrow 23$

$$\begin{split} &w = 1/[\sigma^2(F_o{}^2) + (0.043P)^2 \\ &+ 0.6857P] \\ &where \ P = (F_o{}^2 + 2F_c{}^2)/3 \\ (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.38 \ {\rm e} \ {\rm \AA}{}^{-3} \\ \Delta\rho_{\rm min} = -0.22 \ {\rm e} \ {\rm \AA}{}^{-3} \end{split}$$

Methyl H atoms were placed in calculated positions, with a C–H bond length of 0.98 Å, and have been refined as riding, with $U_{iso}(H) = 1.5U_{eq}(C)$. All other H atoms were located in a difference density Fourier map, and their atomic coordinates were refined. For the acetylenic H atom, the isotropic displacement parameter was also refined; for all other H atoms, $U_{iso}(H) = 1.5U_{eq}(C)$. In the absence of significant anomalous dispersion effects, Friedel pairs were merged before refinement. The absolute configuration assignment is based on the known configuration of C atoms retaining their configuration during the synthesis of the alkyne. The s.u. values of the cell parameters are taken from the software, recognizing that the values are unreasonably small (Herbstein, 2000).





The molecular structure of (I), showing 50% probability displacement ellipsoids.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT-Plus* (Bruker, 2003); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

MZ was supported by NSF grant 0111511, DPT and PN by NIH grant R15 AI053112-01 and the diffractometer was funded by NSF grant 0087210, by Ohio Board of Regents grant CAP-491, and by YSU.

References

Akula, R. A., Temelkoff, D. P., Artis, N. D. & Norris, P. (2004). *Heterocycles*, 63. In the press.

- Bruker (2000). SHELXTL. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2002). SMART for WNT/2000. Version 5.630. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2003). SAINT-Plus. Version 6.45. Bruker AXS Inc., Madison, Wisconsin, USA.
- Herbstein, F. H. (2000). Acta Cryst. B56, 547-557.
- Kolb, H. C., Finn, M. G., Sharpless, K. B. (2001). Angew. Chem. Int. Ed. 40, 2004–2021.